## Ultrafast Photochemical Dynamics of a Magnetically Bi-stable Azo-functionalized Ni-Porphyrin Spin Switch

S. Megow<sup>1</sup>, J. Bahrenburg<sup>1</sup>, H. Böhnke<sup>1</sup>, M. Bohnsack<sup>1</sup>, M. Dittner<sup>1</sup>, D. Bank,<sup>1</sup> M. Dommaschk<sup>2</sup>, C. Schütt<sup>2</sup>, R. Herges<sup>2</sup>, F. Temps<sup>1\*</sup>

<sup>1</sup>Institute of Physical Chemistry, Christian-Albrechts-University, Kiel, Germany <sup>2</sup>Institute of Organic Chemistry, Christian-Albrechts-University, Kiel, Germany

The azopyridine-functionalized Ni-porphyrin **1** (Figure 1) shows magnetic bi-stability in homogeneous solution at room temperature [1]. On irradiation at  $\lambda \sim 435$  nm, the four-coordinate low-spin Ni complex switches to its high-spin state, which is trapped by axial coordination of the azopyridine ligand that undergoes simultaneous *trans*-to-*cis* isomerization. On irradiation at  $\lambda \sim 500$  nm, this process is reversed and the low-spin state is efficiently recovered.

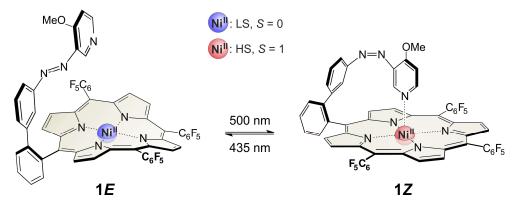


Figure 1: Structures of the azo-functionalized Ni-porphyrin spin switch states.

To gain insight into the ensuing photoswitching mechanisms, we have investigated the ultrafast dynamics of the E (low spin) and Z (high spin) states of 1 using femtosecond transient absorption spectroscopy after excitation in the Q and B bands of the porphyrin. The four-coordinate low-spin species was found to exhibit virtually identical dynamics as the plain Ni-porphyrin complex. Transient spin state switching happens in <0.25 ps, but a stabilization to the long-lived high spin product is very inefficient, making this channel a very minor one. In contrast, the five-coordinate high-spin compound shows virtually immediate (<0.25 ps) formation of the low-spin product, subsequent vibrational cooling in ~12 ps, and persistent positive low-spin product absorption at 1 ns. The high-spin-to-low-spin quantum yield is ~ 6 ± 1 %. The photoinduced dynamics are dominated by coupling of the porphyrin- $\pi\pi^*$  and Ni(d-d)/Ni(d²) states, not by initial photoisomerization of the azopyridine. The ligand cis-trans isomerization may instead take place in the azopyridine triplet state reached via the porphyrin triplet.

## References:

[1] S. Venkataramani, U. Jana, M. Dommaschk, F.D. Sönnichsen, F. Tuczek, R. Herges, Science **331**, 445 (2011).